

The background features a close-up of a thermometer with a red liquid column and a scale marked in degrees Celsius. The numbers 0, 1, and 2 are visible on the scale. In the bottom right corner, there is a digital display showing the time 10:25 and the date 5.08.20 in red. A white speaker icon is located in the bottom right corner.

Unit and Measurement

Unit 1 – Science Knowledge

Learn Objectives

- Know different measurement and know how to use them
- Understand the SI System
 - SI Base Units
 - Meter, kg, L, mL, etc
- Unit Conversion
- Know significant figures
- Know and understand different types of errors
- Know how to calculate average value and uncertainty

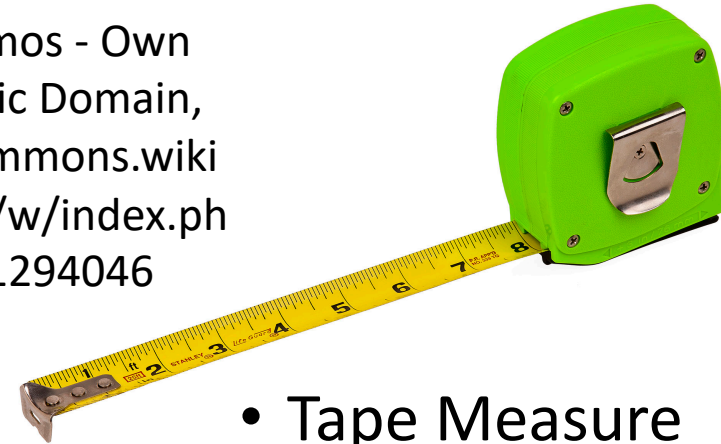


Measurement Tools

- Micrometer

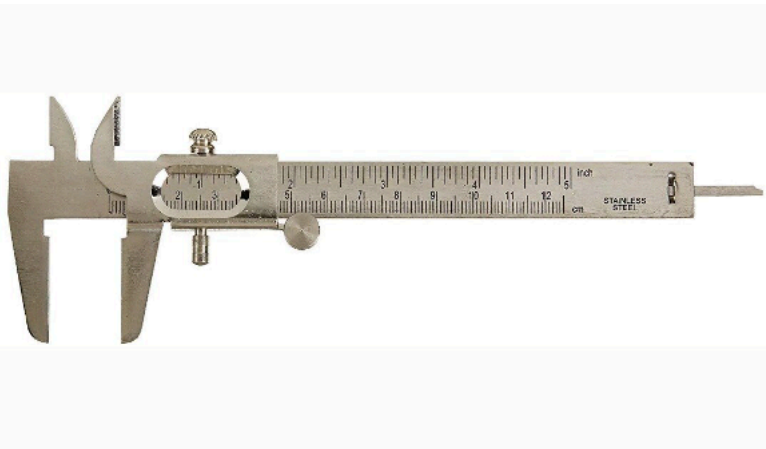


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- Tape Measure

vernier caliper

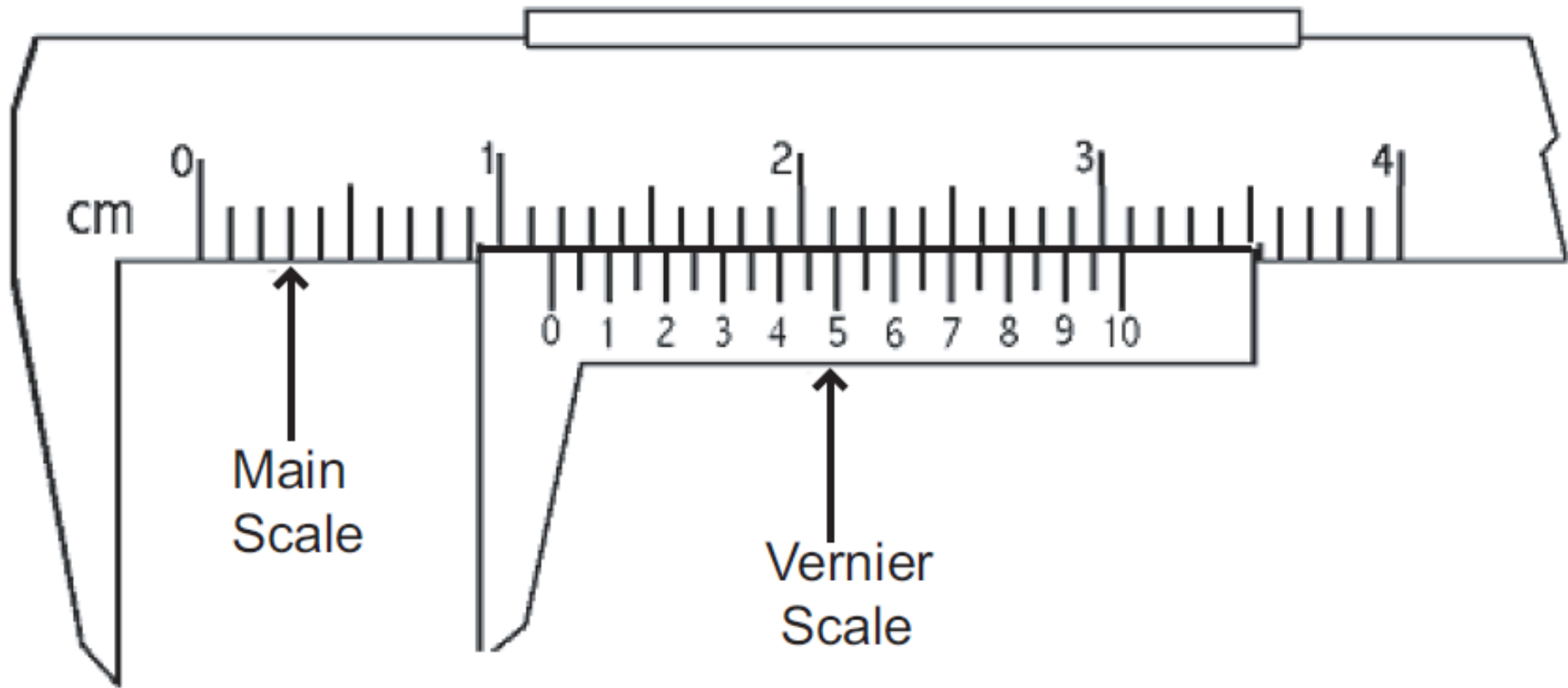


- Tape Measure

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Vernier Caliper



Read a Vernier Caliper

- Follow these steps to read the vernier scale:
- Read the main scale. (Last whole increment visible before the 0 (zero) mark)
- Read the secondary scale measurement. (Division that lines up best with a mark on the main scale.)
- Add the two measurements together.



Significant Figures

- The number of figits used to express a number carries information about how precise the number is known



Counting Significant Number

Number	Significant number	Why?	Scientific Motation
786	3	In a integer, all digits count if the last digit is not zero.	7.86×10^2
859000	3	Zero at the en does not count because its reliability is uncertainty. For example $859999.999999 = 859000$	8.59×10^5
100	1	Zero at the en does not count because its reliability is uncertainty.	1×10^2
0.004900	4	Zero at the end count, those at the front does not count	4.9×10^{-3}
8.38485×10^5	6	Cuase 8.38485 has six significant digits	8.38485



Uncertainty and Errors

Types of Uncertainty

- Systematic and random

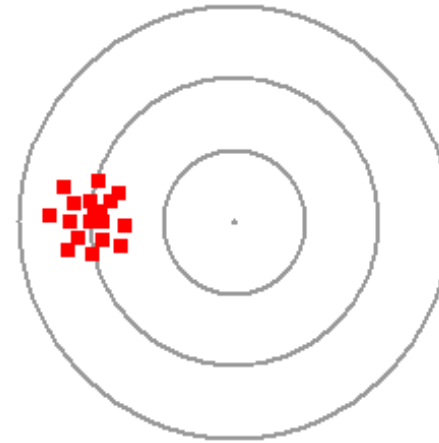
Systematic Error: A systematic is a consistent, repeatable error

- E.g: If you use a ruler to measure the length of a book, the ruler itself might contain some measuring inaccuracy

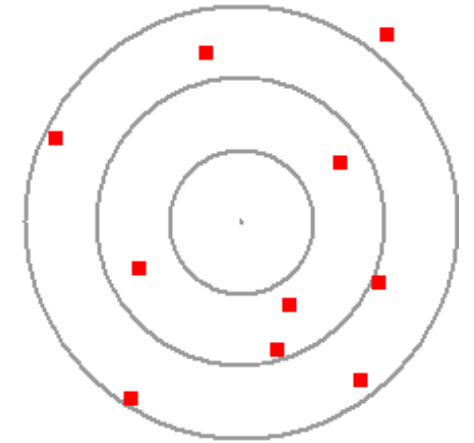


Random Error

- **Random Error**: Error caused by unknown and unpredictable changes in the experiment. These changes may occur in the measuring instruments or in the environmental conditions.
- E.g: You record the amount of time for a student to run 100 meters, but you might start the timer earlier or later, causing random error.



Systematic Error



Random Error



Reduce Random Error

- Doing repeated measurement (Multiple trials of experiment)
- Take an average value of multiples trails of data to get a more **accurate estimate of the result.**
- Calculate uncertainty: **reading uncertainty = \pm half of its samllest precision**
 - There might be reading uncertainty when you are measuring
 - **Precision refers to the closeness of two or more measurements to each other**



Reading Uncertainty

- **Best estimate \pm uncertainty**

- Always round the experimental measurement or result to the same decimal place as the uncertainty

- For Example

- 54.6mL \pm 0.05ml is **wrong**
- 54.60mL \pm 0.05ml is **correct**

- **Wrong:** 1.237 s \pm 0.1 s

- **Correct:** 1.2 s \pm 0.1 s



Reading Uncertainty Example

- The smallest precision of the beaker is 25mL. The uncertainty is $25\text{mL}/2 = 12.25\text{ mL}$
- The volum of the liquid is somewhere between 175mL and 200 mL.
- The average value = $(175 + 200)/2 = 187.50\text{ mL}$
- Reading with uncertainty:
 $187.50\text{mL} \pm 12.25\text{mL}$

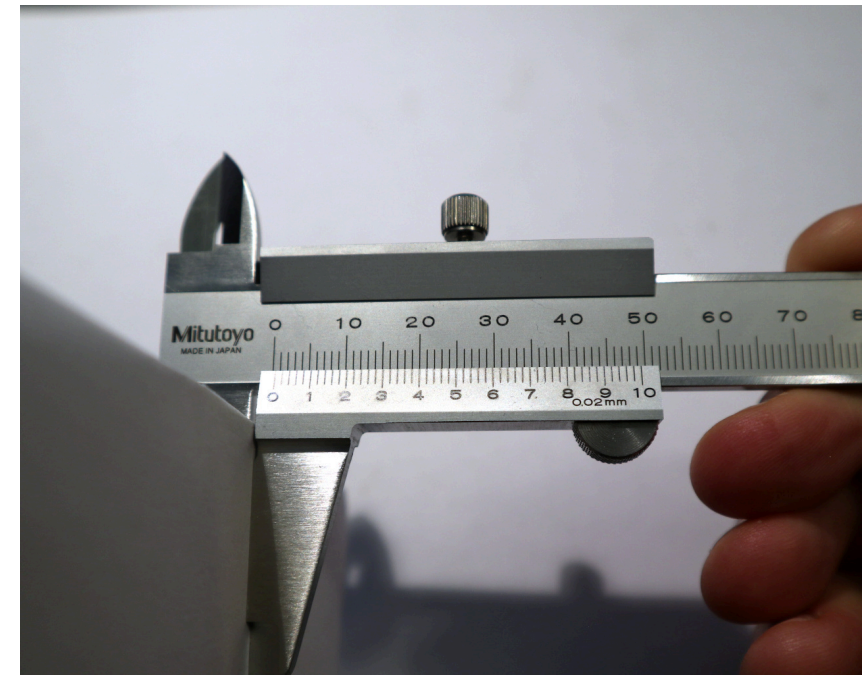


Uncertainty Table

Usually

- Ruler: $\pm 0.05\text{mm}$
- Vernier calipers: $\pm 0.05\text{mm}$
- Micrometer: $\pm 0.005\text{mm}$

- The above measuring tools' uncertainties might change
- The table only shows the frequent cases



Accuracy And Precision

- **Accuracy**: How close to the true/theoretical value
- **Precision**: is the closeness of the measurements to each other.



High accuracy
High precision



Low accuracy
High precision



High accuracy
Low precision



Low accuracy
Low precision



Average, Uncertainty

- $\frac{\text{sum of all terms}}{\text{number of terms}} = \frac{x_1+x_2+x_3+\dots+x_N}{N} = \bar{x}$
- The average is the best estimate for the quantity x based on N measurement.
- $\Delta x = \frac{x_{max}-x_{min}}{2}$



Example 1

Length of a string are listed below

- 163,173,165, 164, 124, 178, 195, 187, 175, 174 (in cm)

- Average Value = $\frac{163+173+165+164+124+178+195+187+175+174}{10} = 169.8\text{cm}$

- Uncertainty = $\frac{195-124}{2} = 35.5\text{cm}$

- **Result: $169.8\text{cm} \pm 35.5 \text{ cm}$**



Calculate the average value and uncertainty

Example 2

Pressure (Pa)	Pressure (Pa)	Pressure (Pa)
101504.0	101560.3	101567.1
101459.0	101541.0	101527.0
101435.6	101511.6	101511.6
101416.8	101493.1	101492.6
101396.6	101472.2	101471.8
101377.6	101454.0	101455.3
101356.5	101428.7	101431.6
101334.1	101413.6	101410.4
101319.5	101392.1	101389.6
101296.7	101367.1	101374.0



Example 2

Pressure	Pressure	Pressure	Max Pressure	Min Pressure	Pressure Uncertainty	Average Pressure
101504.0	101560.3	101567.1	101567.2	101503.9	31.6	101535.5
101459.0	101541.0	101527.0	101541.0	101459.0	41.0	101500.0
101435.6	101511.6	101511.6	101511.6	101435.6	38.0	101473.6
101416.8	101493.1	101492.6	101493.1	101416.8	38.1	101455.0
101396.6	101472.2	101471.8	101472.2	101396.6	37.8	101434.4
101377.6	101454.0	101455.3	101455.3	101377.6	38.9	101416.4
101356.5	101428.7	101431.6	101431.6	101356.4	37.6	101394.0
101334.1	101413.6	101410.4	101413.7	101334.1	39.8	101373.9
101319.5	101392.1	101389.6	101392.1	101319.5	36.3	101355.8
101296.7	101367.1	101374.0	101374.0	101296.7	38.7	101335.4



Propagation of Uncertainties

- $28.3\text{cm} \pm 0.4\text{ cm}$

- 28.3 cm is the best estimate/mean value/average value, and 0.4 cm is the absolute uncertainty

- $\frac{\textit{absolute value}}{\textit{mean value}} = \textit{fractional uncertainty} = \frac{0.4}{28.3} = 0.0141$

- Percentage uncertainty = $0.0141 \times 100\% = 1.41\%$



Propagation of Uncertainties

In General

- Absolute uncertainty = Δa
- Fractional uncertainty = $\frac{\Delta a}{a}$
- Percentage uncertainty = $\frac{\Delta a}{a} \times 100\%$

$$a \pm \Delta a$$

Best estimate \pm uncertainty



Uncertainty-Addition and Subtraction

- Calculate the estimate values, then \pm all absolute uncertainty

Example

- $A = 3.56 \pm 0.23$
- $B = 193.34 \pm 2.20$
- $A + B = (3.56 + 193.34) \pm (0.23 + 2.20)$
- $B - A = (193.34 - 3.56) \pm (0.23 + 2.20)$



Multiplication and Division

- $Q = ab \rightarrow \frac{\Delta Q}{Q_o} = \frac{\Delta a}{a_o} + \frac{\Delta b}{b_o}$
- $Q = \frac{a}{b} \rightarrow \frac{\Delta Q}{Q_o} = \frac{\Delta a}{a_o} + \frac{\Delta b}{b_o}$
- $Q = \frac{ab}{c} \rightarrow \frac{\Delta Q}{Q_o} = \frac{\Delta a}{a_o} + \frac{\Delta b}{b_o} + \frac{\Delta c}{c_o}$



Work Example

- The side of a rectangle is measured to be $a = 2.5\text{cm} \pm 0.1\text{cm}$, and $b = 5.0\text{cm} \pm 0.1\text{cm}$. Find the area A of the rectangle



Work Example

- The side of a rectangle is measured to be $a = 2.5\text{cm} \pm 0.1\text{cm}$, and $b = 5.0\text{cm} \pm 0.1\text{cm}$. Find the area A of the rectangle

- $\frac{\Delta a}{a_o} = \frac{0.1}{2.5} = 0.04$ or 4%

- $\frac{\Delta b}{b_o} = \frac{0.1}{5.0} = 0.02$ or 2%

- $\frac{\Delta A}{A_o} = 0.04 + 0.02 = 0.06$

- $A_o = (2.5 \times 5.0) = 12.5 \text{ cm}^2$

- $\Delta A = (0.06) \times (12.5) = 0.75 \text{ cm}^2$

Final Answer

$$12.5 \text{ cm}^2 \pm 0.75 \text{ cm}^2$$



Powers and Roots

- $Q = a^b \rightarrow \frac{\Delta Q}{Q_o} = |b| \frac{\Delta a}{a_o}$
- $Q = \sqrt[n]{a} \rightarrow \frac{\Delta Q}{Q_o} = \frac{1}{|n|} \times \frac{\Delta a}{a_o}$



Worked Example

- The length of a simple pendulum L is increased by 4%. What is the fractional increase in pendulum's period T . Given that $T=2\pi\sqrt{\frac{L}{g}}$



Worked Example

- The length of a simple pendulum L is increased by 4%. What is the fractional increase in pendulum's period T . Given that $T=2\pi\sqrt{\frac{L}{g}}$
- $\frac{\Delta T}{T_0} = \frac{1}{2} \times \frac{\Delta L}{L_0} = \frac{1}{2} \times 4\% = 2\%$
- The fractional increase in pendulum's period T is 2%



References

- https://www.123rf.com/photo_17309282_best-icon-located-on-a-white-background.html
- Physics for the IB Diploma, six edition ny K.A Tsokos (Cambridge University Press)
- <https://www.amazon.in/SSU-Silver-Vernier-Caliper-Range/dp/B01N8UUMKH>
- https://en.wikipedia.org/wiki/Tape_measure
- https://www.stefanelli.eng.br/en/virtual-vernier-caliper-simulator-05-millimeter/#swiffycontainer_2

